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Antituberculosis Activity of Natural and Synthetic Compounds

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Abstract

The review presents the most active natural and synthetic compounds those exhibit antimycobacterial activity providing the minimum inhibiting concentration (MIC) $\leq 5~\mu g/mL$. For better understanding the structure–activity relationship, compounds with a high value of MIC are considered in some cases. The review covers the papers published within the range of 2001–first half of 2009. The information in the review is systematized with respect to chemical structures (the nitrogen-, oxygen-, sulphur-containing heterocyclic compounds, peptides, alkaloids, terpenoids and others.).

Key words: tuberculosis, tuberculosis with plural drug resistance, *M. tuberculosis*, antimycobacterial activity, antituberculosis activity, cytotoxicity, structure–property relationship

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INTRODUCTION

Tuberculosis represents a chronic infectious disease, it is known since extreme antiquity. This disease remains the most large-scale problem not only from medical, but also from social viewpoint for the present day, too. Annually, owing to tuberculosis about 3 million people dies all over the world and approximately 8 million events of first registered tuberculosis are observed every year. The progress in tuberculosis chemotherapy in the middle of 20th century was in the recent time changed

into solicitude about the evolution of drug resistance on the base of genetically fixed mutation of *M. tuberculosis*.

Besides, almost all medicinal preparations used for the treatment of tuberculosis those exhibit different mechanisms of the action, are capable of exerting negative side effects on a human organism. In this connection, the problem of searching for new, low toxic medicinal agents, exceeding the existing preparations in the activity and efficiency is extremely urgent. First of all, all this concerns the agents efficient in combating against strains of *M. tuber*-

culosis with plural resistance with respect to pharmaceutical preparations.

Contemporary tuberculosis is usually connected with M. tuberculosis and M. bovis, the mycobacteria pathogenic for human beings. Owing to slow growing and high pathogenicity of *M. tuberculosis* H37Rv many research groups are using fast-growing and/or nonpathogenic mycobacteria, including M. tuberculosis H37Ra, M. smegmatis, M. aurum and others as an organism under testing. Besides, antimycobacterial activity is studied also using such strains as M. avium and M. intracellular those cause bird tuberculosis associating with human diseases in developed countries (AIDS patients and immunocompromised individuals), for the purpose of finding the compounds those exhibit a broad spectrum of activity. Studies should be especially noted performed with clinical isolates and strains of M. tuberculosis those exhibit plural resistance with respect to pharmaceutical preparations. Tuberculosis with plural medicinal resistivity (multidrug resistant TB, MDRTB) is unequivocally determined as the strain of M. tuberculosis resistant simultaneously against both isoniazid and rifampicin [1, 2]. Tuberculosis with other drug resistance (ODRTB) corresponds to the strains of M. tuberculosis those exhibit mono- or poly-resistance which includes combined resistance with respect to isoniazid and rifampicin [3]. Researchers distinguish susceptible M. tuberculosis strains (those can be inhibited by the preparations of the first series, for instance, by isoniazid) and resistant M. tuberculosis strains (those are not inhibited by isoniazid). For the work, and/or organisms under testing are used different methods of analysis, which is necessary for taking into account in comparing biological activity values obtained from different studies.

As reference preparations when testing the compounds, researchers often use preparations applied in the modern medicinal therapy of tuberculosis. There are several categorizations of antituberculosis medicinal preparations. So, medicinal preparations are divided into basic remedies (isoniazid, rifampicin, pyrazinamide, and ethambutol) and reserve remedies (cycloserine, kanamycin, ethionamide). The other way to divide into groups the medicinal preparations consists in taking into account the effi-

ciency. The I group/series (the most efficient medicinal preparations) includes isoniazid and rifampicin, the II group (the preparations of medium efficiency) includes ethambutol, streptomycin, ethionamide, pyrazinamide, kanamycin, cycloserine), the III group (the preparations with moderate activity) includes PASA (para-aminosalicylic acid), thioacetazone [4].

The presented review covers the papers within the period since 2001 up to the first half of 2009, whereas the selected structures are characterized by values of minimal inhibition concentration (MIC) $\leq 5~\mu g/mL$. Such a restriction allowed the authors to gather and analyze the most efficient compounds in one review. In some cases, for better understanding the structure–property relationship, structures with higher MIC values are presented. In order to separate any possible structure–activity relations, the information in the review is systematized taking into account chemical structure.

SYNTHETIC COMPOUNDS WITH ANTIMYCOBACTERIAL ACTIVITY

Nitrogen-containing heterocycles

Chiral pentaamines 1 (MIC = $3.13 \mu g/mL$), bicyclic guanidines 2 (MIC = $3.9 \mu g/mL$) and piperazines 3 (MIC = $2-3.9 \mu g/mL$) inhibit the growing of M. tuberculosis H37Rv (ATCC 27294) more efficiently by contrast with ethambutol (MIC = $10 \mu g/mL$) [5] (Scheme 1).

For the present day, 1,5-diarylpyrroles are well studied those represent the analogues of compound BM 212 (MIC 1 µg/mL), the leader structure of this class of compounds and one of the most promising antimycobacterial preparations [6]. Pharmacophore ligands were determined for them; the chemical groups in the pyrrole inhibit the growing ring determining a high level of activity. The key moment for the occurrence of antituberculosis activity of 1,5diarylpyrroles consists in the presence of (thiomorpholine-4-yl)methyl fragment at C-3 atom of the pyrrole ring, since N-methylpi perazine methyl derivatives are much more toxic and much more active than corresponding methyl thiomorpholine compounds. The improvement of antituberculosis activity thiomorpholine methyl derivatives is promoted by the introduc-

Scheme 1.

tion of a halophenyl substituent into the molecule (in this case, a strong effect is exerted by the nature of halogen atom), as well as the presence of methyl group in the second phenyl ring much more lipophilic, than the atom of halogen.

Several novel compounds (4a-c), MIC = 1, 0.4 and 0.5 µg/mL, respectively) were found, whose activity with respect to M. tuberculosis is comparable with isoniazid, streptomycin and rifampicin. Additionally, they exhibit much more low cytotoxicity by the contrast with isoniazid and streptomycin comparable with the cytotoxicity value observed for rifampicin. It should be noted that compound 4a-c are active against intracellular (intramacrophage) M. tuberculosis, herewith the MIC value of compound 4a is three times lower comparing to the MIC value of rifampicin (1 and 3 µg/mL, respectively). In addition, compounds 4a-c are active against different strains and clinical isolates of M. tuberculosis resistant with respect to one or simultaneously against several medicin al preparations [7, 8] (Scheme 2).

The influence of different substituents in aromatic rings of 1,5-diarylpyrroles was studied concerning the activity against M. tuberculosis, atypical mycobacteria and resistant strains. As the M. tuberculosis 103471 is concerned, compounds 4d-t are active alongside with isoniazid, streptomycin and rifampicin, except for compound 4i (MIC = $16~\mu g/mL$). So, for compound 4o and isoniazid MIC = $0.125~\mu g/mL$,

for compounds 4d,g,l,m,r,s, and rifampicin MIC = $0.25~\mu g/mL$, whereas for compounds 4k,o,p and streptomycin MIC = $0.5~\mu g/L$. In the case of rifampicin-resistant strain (the MIC value for rifampicin is higher than $64~\mu g/mL$) the order changes: compounds 4m,n exhibit the MIC value amounting to $0.125~\mu g/mL$, for compounds 3~g,l,p,r and isoniazid MIC= $0.25~\mu g/mL$, whereas for streptomycin MIC= $4~\mu g/mL$ [6].

As the result of the further structural optimization of this class of compounds researchers found the compound $4t~(\mathrm{MIC}=0.25~\mu\mathrm{g/mL}$ for M. tuberculosis CIP 103471, M. tuberculosis H37Rv ATCC 27294 and the rifampicin-resistant M. tuberculosis ATCC 35838; very low cytotoxicity) which compound exhibits a better biological profile than its 2-methyl analogue 4b and reference preparations such as streptomy-

Scheme 2.

cin (MIC = $0.5~\mu g/mL$ for three strains) and rifampicin (MIC= $0.25~\mu g/mL$ for two strains). Other 1,5-diaryl-2-ethyl-pyrroles were active, too (MIC = $1-4~\mu g/mL$) being more efficient than 2-methyl analogues of their own [9].

It should be noted that the derivatives of pyrrole presented here exhibit a high selectivity with respect to *M. tuberculosis* being inactive against atypical mycobacteria *M. gordonae* 6427, *M. smegmatis* 103599, *M. marinum* 6423 and *M. avium* 103317 (mainly, MIC > 16 µg/mL).

The best results concerning the inhibition of growing M. tuberculosis H37Rv have demonstrated only two compounds (5a,b, MIC = $4 \mu g/$ mL) from two series of novel pyrozolone derivatives. However basing on structure-activity correlation studied the authors of [10] have drawn an important conclusion: for the antituberculosis activity of a compound, the presence of para-chlorobenzene fragment at C-4 atom of the pyrazole ring is required. Six derivatives of imidazole 5c well inhibit the growing of M. tuberculosis H37Rv (MIC = 4 $\mu g/mL$, for the comparison (the MIC value for isoniazid amounts to 0.5 µg/mL. The presence of phenyl or 4-fluorophenyl substituent at N-1 atom of the pyrazole ring appeared much more preferable than the methyl group or hydrogen atom in this position [11] (Scheme 3).

Oxazolidinones represent a novel class of synthetic antimicrobial remedies those exhibit no cross-resistance with respect to the other

5a,b R = N-Me-piperazine, morpholine

$$R \xrightarrow{N \atop N-N} R$$

5c R = 4-Br, 4-Cl, $2,4(Cl)_2$, 4-Me $R^1 = Ph$, 4-F-Ph

Scheme 3.

types antibiotics, since a new mechanism of action is inherent in them. Amongst them there is linezolid (MIC = $0.25-2 \mu g/mL$ for M. tuberculosis H37Rv, of susceptible and resistant clinical isolates), and its thiomorpholine analogue PNU-100480 6a exhibiting an interesting antituberculosis activity. Continuing the work concerning the development of antituberculosis preparations a new series of 1-[3-(4-benzotriazole -1/2-yl-3-fluorophenyl)-2-oxo-oxazolidine-5-yl-methyl]-3 derivatives of thiourea was created. A complete activity loss was caused by the substitution of the hydrogen atom in the terminal ethyl group of thiourea by the morpholine fragment. A significant activity against different mycobacterial species (M. tuberculosis H37Rv, susceptible and resistant clinical isolates) is exhibited by compound **6b-f** (MIC = $0.5-8 \mu g$ / mL), having amino-, 2-pyridyl-, 1-pyrrolidinyland 1-piperidinyl groups bound by ethyl bridge with thiourea. Substituting the whole ethyl group by the cyclopropyl one results in the forming the compound 6d which exhibits an excellent antituberculosis activity (MIC = $0.06-2 \mu g/mL$), comparable with the activity of linezolid with exceeding the activity of isoniazid for all the tested strains [12]. Amongst different C-5 triazole substituted oxazolidinones there is 3-(4acetylphenyl)-5-(1H-1,2,3-triazole-1-yl)methyl)oxazolidinone $\mathbf{6g}$ (MIC = 1 $\mu g/mL$ with respect to M. smegmatis ATCC 14468), fourfold exceeding the activity of isoniazid [13].

A high inhibiting activity of peptide deformy-lase for mono- and poly-resistant strains of M. *tuberculosis* was demonstrated by compound **7a-e** (MIC = $0.03-0.6 \mu g/mL$) [14] (Scheme 4).

A stage-by-stage modification was successfully performed for well known bicyclic nitroimidazooxazole **8** which exhibit not only satisfactory *in vitro* antituberculosis activity and *in vivo* efficiency, but also mutagenicity. Compounds those exhibit no mutagenicity were revealed among the structures containing heteroatomic substituents at the 2nd position of 6-nitro-2,3-dihydroimidazo[2,1-b]oxazole, herewith (R)-form **8a** appeared much more active (MIC = 0.05 µg/mL with respect to M. tuberculosis H37Rv).

The improvement of antituberculosis activity was attained by means of consequent introducing the molecule with a hydrophilic substituent (compound 8b, MIC = $0.78 \mu g/mL$ with

 $R = NH_2, NHCO-4-pyridyl, CH(Et)CH_2OH, \\ CH_2CH(OEt)_2, CH_2CH_2NH_2 (6b), CH_2CH_2-2-pyridyl (6c), \\ cyclopropyl (6d), CH_2CH_2-pyrrolidine (6e), \\ CH_2CH_2-piperidine (6f), CH_2CH_2-morpholine (6g)$

NHR

Scheme 4.

respect to M. tuberculosis H37Ra and MIC = 0.39 µg/mL with respect to each of two monoresistant strains of M. tuberculosis H37Rv (mono-resistance against isoniazid and against rifampicin, respectively) and with a lipophilic substituent (compound 8c, MIC = $0.006 \mu g/mL$ with respect to each of two mono-resistant strains of M. tuberculosis H37Rv (mono-resistance against isoniazid and rifampicin, respectively). Basing on the excellent in vitro antituberculosis activity with respect to drug-susceptible and drug-resistant strains of M. tuberculosis H37Rv as well as in vivo efficiency in mice infected by M. tuberculosis Kurono, compound 8c was chosen as an active remedy nomin ated for oral introduction in tuberculosis treatment. Particularly significant is the fact that compound 8c in vivo is highly competitive in the efficiency with rifampicin at a much lower dose for oral introduction (0.313 and 5 mg/kg, respectively) [15] (Scheme 5).

The oxyindole derivatives **9a**, **9c** (MIC = 0.1 μ g/mL for each) and **9b** (MIC = 0.05 μ g/mL)

7b R = Bu, Y = NH 7c R = Cpm, Y = NH 7d R = Bu, Y = O 7e R = Cpm, Y = NH

$$O_2N$$
 N
 O_2N
 N
 O_2N
 N
 O_2N
 N
 O_2N
 N
 O_2N
 O

Scheme 5.

exhibit the antituberculosis activity with respect to $M.\ tuberculosis\ H37Rv$, comparable with the effect of isoniazid (MIC = $0.025-0.2\ \mu g/mL$) and rifampicin (MIC $0.06-0.5\ \mu g/mL$) [16].

For the ability to inhibit M. tuberculosis H37Rv (ATTC 27294) a series of phthalimide derivatives was tested. Among the compounds studied, a good antituberculosis activity was demonstrated by compound 10a, b (MIC = 3.9 and 5 $\mu g/mL$, respectively), which alongside with the absence of cytotoxicity allows one to choose these compound as leading structures for obtaining more efficient antituberculosis remedies [17].

Studies were performed concerning the antituberculosis activity for the series of hetarylamides of 1-hydroxy-3-oxo-5,6-dihydro-3*H*-pyrrolo[3,2,1-*ij*]quinoline-2-carboxylic and 1-hydroxy-3-oxo-6,7-dihydro-3*H*,5*H*-pyrido[3,2,1-*ij*]quinoline-2-carboxylic acids, basing on which some regularities in biological structures were revealed. A pronounced antituberculosis effect with respect to *M. tuberculosis* H37Rv ATCC 27294 is exhibited by compound (11a-l, MIC < 3.13 µg/mL), from those the five of the most promising compounds were chosen (11g-k, MIC = 0.39-0.78 µg/mL) [18], and compounds (11m-o,s, with MIC = 3.13

 μ g/mL) and (11p,r, with MIC= 0.78 and 1.56 μ g/mL, respectively) [19] (Scheme 6).

The studies on 3,5-bis(benzylidene)-4-pi peridones* and their N-4-(2-aminoethoxy)phenyl carbonyl analogues 12a has allowed researchers to reveal a number of substituents determining not only high antituberculosis characteristics (MIC = $1.56-3.13~\mu g/mL$), but also the absence of neurotoxicity. The leading structures of this class of compounds are presented by compound 12b,c (MIC = 0.2 and $0.78~\mu g/mL$, respectively) [20].

Novel spiroheterocycles 12d-f (MIC = $0.4-3.13\,\mu g/mL$) obtained on the base of 1-methyl-3,5-bis(benzylidene)-4(1H)-pi peridone, highly active not only against $M.\ tuberculosis\ H37Rv$, but also against clinical isolates with MDRTB resistant with respect to isoniazid, rifampicin, ethambutol and ci profloxacin [21].

Among 4-substituted imidazolyl-2,6-dimethyl- N^3 , N^5 -bisaryl-1,4-dihydropyridine-3,5-dicarboxamides, the most pronounced activity is

*In the Schemes 7–38 the imaging of substituents near the frame under discussion without labelling the particular number of a compound means that for the structure presented MIC > 5 $\mu g/mL$.

Scheme 6.

antituberculosis activity:
$$R = H, Me$$

$$amine = -NMe_2, -N$$

$$O_2 = N$$

$$O_2 = N$$

$$O_3 = N$$

$$O_4 = N$$

$$O_4$$

13a $R^1 = 4$ -Cl-Ph, $R^2 = Bz$ **13b** $R^1 = 3$ -pyridyl, $R^2 = Bz$

Scheme 7.

exhibited compound **13a** (MIC = 1 μ g/mL for *M. tuberculosis* H37Rv ATCC 27294) which is an equivalent with respect to the action of reference preparation rifampicin. A good inhibiting ability is exhibited also by analogue **13b** (MIC = 2 μ g/mL for *M. tuberculosis* H37Rv ATCC 27294). Compounds with other substituent (R¹= 2-Cl-Ph, 3-Cl-Ph, 4-Br-Ph, 2-pyridyl; R² = NHPh) represent much more weak inhibitors [22] (Scheme 7).

Novel efficient inhibitors for P450 14α -sterol demethylase of M. tuberculosis were revealed such as 3.5-disubstituted derivatives 1,3,4-oxadiazole-2(3H)-one 14a-k (MIC = 4 μ g/mL) [23] (Scheme 8).

Compounds **15a** obtained by means of isoniazid structure modification at N-2 atom exhibit a high level of antituberculosis activity (*M. tuberculosis* H37Rv and *M. tuberculosis* Erdman *in vitro* and *in vivo* in experimental ani-

Scheme 8.

$$R^{1} = (CH_{2})_{4}CH_{3}, \ H, \ CH_{2}Ph; \ R^{2} = Ph, \ 4-Cl-C_{6}H_{4}, \ 3,4-Cl_{2}-C_{6}H_{3}, \\ 2,6-F_{2}-C_{6}H_{3}, \ 4-Cl-3-NO_{2}-C_{6}H_{3}, \ CH_{2}CO_{2}CH_{2}CH_{3} \\ R^{1}R^{2} = CCH_{2}CH(CH_{3})CH_{2}C(CH_{3})_{2}CH_{2}, \ C(CH_{2}CH_{2})_{2}CH_{2}$$

Scheme 9.

mals) as well as a low level of toxicity. Among 44 compounds, for 34 ones the typical MIC values are amounting to $0.025-0.5\,\mu g/mL$ (for isoniazid MIC = $0.06\,\mu g/mL$). Eleven compounds tested for infected macrophages exhibit bacteriostatic and bactericidal activity equivalent or even greater than those for isoniazid (EC₉₀ = $0.028-0.106\,\mu g/mL$, EC₉₉ = $0.121-0.5\,9\,6\,\mu g/mL$, respectively; for isoniazid EC₉₀ = $0.03\,\mu g/mL$, EC₉₉ = $0.42\,\mu g/mL$) [24] (Scheme 9).

Excellent antituberculosis characteristics studied *in vivo*, alongside with low cytotoxicity is exhibited by fullerene-isoniazid conjugate **15b** (MIC = 0.5, 2.5 and 5 μ g/mL for *M. tuberculosis* H37Rv ATCC 27294, *M. tuberculosis* H6/99 and *M. avium* ATCC 27291, respectively, the MIC value of isoniazid is 0.25 μ g/mL for *M. tuberculosis* H37Rv ATCC 27294, in the other cases isoniazid is resistant). Arabinoses for obtaining this compound **15c,d** (MIC = 5 μ g/mL for *M. tuberculosis* H37Rv ATCC 27294) also exhibit the antituberculosis activity [25] (Scheme 10).

Isoniazid and pyrazinamide are widely used as basic medicinal preparations for the treatment of tuberculosis, usually together with other drugs. The authors have designed and synthesized a new type antituberculosis compounds those could be considered as «double

Scheme 10.

active» molecules since isoniazid or pyrazinamide are linked therein with another standard antituberculosis drug (ciprofloxacin, para-aminosalicylic acid) by means of C-H group. The compounds were obtained 15e (MIC = $0.39 \,\mu g/mL$), **15f** (MIC = $3.13 \,\mu g/mL$), **15g** (MIC = $0.39 \,\mu g/mL$) mL), 16a (MIC = $0.78 \,\mu g/mL$), 16b (MIC = $3.13 \,\mu\text{g/mL}$), $16c \,(\text{MIC} = 0.1 \,\mu\text{g/mL}) \,\text{exhibit a}$ very high antituberculosis activity with respect to M. tuberculosis H37Rv (the MIC for isoniazid amounting to 0.025-0.2 µg/mL, the MIC for pyrazinamide being equal to 6-60 µg/mL, the MIC for ciprofloxacin is of 2.00 µg/mL), which could be connected with the synergetic action of components and a high lipophilicity of compounds obtained that corresponds to the efficient transport of molecules through cellular membrane [26]. Antituberculosis activity in vitro and in vivo with respect to M. tuberculosis H37Rv and MDRTB strains resistant with respect to isoniazid, rifampicin, pyrazinamide and ofloxacin is exhibited by the aminomethylene amide analogues of the first series pyrazinamide preparation having aryl substituents in the piperazine ring such as compounds 16d (MIC= 3.12 and $12.5 \,\mu g/mL$, respectively), **16e** (MIC = 3.12 and $6.25 \,\mu g/$ mL), **16f** (MIC =1.76 and 1.76 μ g/mL), **16g** (MIC= 0.78 and 1.76 µg/mL). Halogen-containing (Cl, F) substituents in the aryl ring cause the activity to enhance [27] (Scheme 11).

Compounds 17 and 18 (MIC < 0.25 and $0.5 \,\mu\text{g/mL}$, respectively) the most active with respect to M. tuberculosis H37Ra resulted from tested esters of pyrazinoic and quinoxalinic acids, have the 4-acetoxy group in the phenyl ring. It is interesting that comparing to them 2-nitro- and 4-nitrobenzyl analogues exhibit much lower activity. This indicates the fact that under those conditions (pH 6.6), to all appearance, the reduction of the nitro group into amino group does not occur, whereas enzymatic deacylation of 4-acetoxy group can proceed [28].

Scheme 11.

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Novel inhibitors were found with efficient and selective antituberculosis characteristics with respect to M. bovis, M. tuberculosis and M. avium such as acetylene derivatives of 2',3'-dideoxyuridine and 3'-fluoro-2',3'-dideoxyuridine. The most promising representatives of this class of compounds are presented by compounds 19 and 19b, successfully inhibiting the growth of M. bovis, M. tuberculosis (MIC₉₀ = $1-2 \, \mu \text{g/mL}$) and drug-resistant strains of M. tuberculosis [29].

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The activity with respect to M. tuberculosis, M. bovis and M. avium was studied for a number of $1-\beta-D-2'$ -arabinofuranosyl- and $1-(2'-deoxy-2'-fluoro-\beta-D-ibofuranosyl)$ pyrimidine nucleosides with different substituents at C-5 carbon atom of uracyl (alkinyl, alkenyl, alkyl)

and halogen). A high antituberculosis activity among the compounds studied with respect to M. tuberculosis and M. bovis was demonstrated by nucleosides 19c (MIC $_{90} = 1-5 \mu g/mL$), 19d (MIC $_{90} = 1-5 \mu g/mL$) and 19e (MIC $_{90} = 1 \mu g/mL$). The values of MIC $_{90}$ for these compounds are comparable with those for rifampicin (MIC $_{90} = 0.5-1 \mu g/mL$). However, being of the same concentration, these compounds actively inhibit also the growth of a rifampicin-resistant M. tuberculosis H37Rv strain [30] (Scheme 12).

On the base of imidazooxazine PA-824 (MIC = $0.015-0.25 \,\mu g/mL$), the most modern antimycobacterial agent under clinical trials, novel imidazo[1,2-c]pyrimidines 20a-g (MIC = $2-5 \mu g/mL$ with respect to M. tuberculosis H37Rv) were designed. The most low MIC parameter is exhibited by compound 20b (2 µg/ mL), which comparable with the MIC value for reference preparation amikacin (2 µg/mL), however higher comparing to the MIC value for reference preparation rifabutine (0.05 µg/mL). The results of the present investigation indicate the fact that one could obtain compound with a good antimycobacterial potential via bioizosteric substitution of imidazooxazine ring in PA-824 by condensed imidazopyrimidine ring [31] (Scheme 13).

High efficiency, selectivity and low cytotoxicity are inherent in novel quinoxaline derivatives. Their antituberculosis activity depends on the nature of substituent R^1 in the quinoxaline nucleus: the presence of chloro, methyl and methoxy group at the position 7 of the

 $\begin{array}{ll} R = (CH_2)CH_3, & (CH_2)_7CH_3, & (CH_2)_9CH_3 \mbox{ (19a)}, \\ (CH_2)_{10}CH_3, & (CH_2)_{11}CH_3, & C_6H_5 \end{array}$

$$\begin{array}{c} O \\ HN \\ O \\ N \end{array}$$

 $\begin{array}{lll} R = (CH_2)_4 CH_3, & (CH_2)_7 CH_3, & (CH_2)_9 CH_3, \\ (CH_2)_{10} CH_3, & (CH_2)_{11} CH_3 & \textbf{(19b)}, & 4\text{-Pr-}C_6 H_4 \end{array}$

HO O HO F

19c

$$R = C_{10}H_{21}$$
 $R = C_{10}H_{21}$
 $R = C_{10}H_{21}$

Scheme 12.

$$F_{3}C$$

$$O$$

$$PA-824$$

$$EtO_{2}C$$

$$HN$$

$$R^{2}$$

$$R^{1}$$

 $\begin{array}{c} \textbf{20b} \ R^1 = 4\text{-MeO-C}_6 H_4 \\ R^2 = H \end{array}$

Scheme 13.

benzene ring causes the minimum inhibiting concentration to decrease. However, the activity of these compounds depends also on the substituent R⁴ in the following order: benzyl > ethyl > 2-methoxyethyl > allyl > tert-butyl. The studies concerning macrophages infected with tuberculosis, allowed revealing compounds 21a-d with a good antituberculosis by activity (MIC =1.56, 0.20, 0.10, 0.10 µg/mL, respectively). Compound 21e is active with respect to seven different monoresistant strains (MIC = 0.39-1.56 µg/mL) [32].

For 34 compounds with general formula **21f**, among 70 tested 3-phenylquinoxaline-1,4-di-N-oxides the MIC value is less than $2 \,\mu g/mL$, which is comparable with the activity of rifampicin (MIC = $0.125 \,\mu g/mL$) [33].

More efficiently comparing to rifampicin (reference substance, MIC = $1.00 \,\mu\text{g/mL}$), $M.\,tuberculosis\,H37Rv$ (ATCC 27294) is inhibited by and macrolactone **21g** (MIC = $0.62 \,\mu\text{g/mL}$) [34] (Scheme 14).

Though phenoxazines, phenothiazines and acridines are well known pharmacophores connected with antituberculosis activity, the antituberculosis profile was for the first time reported concerning 1,2,3,4-tetrahydroacridines those have different substituents at C-9 atom. Among 9-aminoalkyltetrahydroacridines, compound 22a (MIC = $1.56 \,\mu\text{g/mL}$) is much more specific with respect to an avirulent strain H37Ra, whereas compound 22b is more specific with respect to virulent strain H37Rv (MIC = $0.78 \,\mu g/mL$). Substituting the aminoalkyl group at C-9 atom by phenoxy or thiophenyl group results in the loss of activity, which indicates that nitrogen atom is required for the activity to be exhibited. Only one compound (22c, MIC = $3.12 \,\mu g/mL$) among five bisacridine derivatives appeared active against an avirulent strain. This fact indicates that introducing more than one acridine unit into the molecule does not result in any advantage [35]. Two derivatives of phenazine (23a and 23b, MIC = 1-4 μg/mL) actively inhibit the growth of drugsusceptible and drug-resistant strains of M. tuberculosis. As clinical isolates with drug resistance are concerned, these two compounds demonstrated much better results than isoniazid. The studies demonstrated that the antituberculosis activity of this type of compounds depends on the lengths of the alkyl chain be-

$$R^{1} = H, \text{ Me, OMe, Cl, } F; \\ R^{2} = H, \text{ Me, Cl, } F; \\ R^{3} = \text{Me, CH}_{2}\text{CDCBt} \\ R^{4} = \text{Me, Et, } tert\text{-Bu, } \\ CH_{3}\text{CH}_{2}\text{CH}\text{-CH}_{2}, \text{ CH}_{2}\text{Ph} \\ CH_{2}\text{CH} = \text{R}^{2} = \text{R}^{3} = \text{Me, } R^{4} = \text{Et}^{2} \\ 21e \ R^{1} = R^{2} = H, \ R^{3} = \text{Me, } R^{4} = \text{Et} \\ 21e \ R^{1} = R^{2} = H, \ R^{3} = \text{Me, } R^{4} = \text{CH}_{2}\text{Ph} \\ 21e \ R^{1} = R^{2} = H, \ R^{3} = \text{Me, } R^{4} = \text{CH}_{2}\text{Ph} \\ 21e \ R^{1} = R^{2} = R^{3} = \text{Me, } R^{4} = \text{Et}^{2} \\ 21e \ R^{1} = R^{2} = R^{3} = \text{Me, } R^{4} = \text{Et}^{2} \\ 21e \ R^{1} = R^{2} = R^{3} = \text{Me, } R^{4} = \text{Et}^{2} \\ 21e \ R^{1} = R^{2} = R^{3} = \text{Me, } R^{4} = \text{Et}^{2} \\ 21e \ R^{1} = R^{2} = R^{3} = \text{Me, } R^{4} = \text{Et}^{2} \\ 21e \ R^{1} = R^{2} = R^{3} = \text{Me, } R^{4} = \text{Et}^{2} \\ 21e \ R^{1} = R^{2} = R^{3} = \text{Me, } R^{4} = \text{Et}^{2} \\ 21e \ R^{1} = R^{2} = R^{3} = \text{Me, } R^{4} = \text{Et}^{2} \\ 21e \ R^{1} = R^{2} = R^{3} = \text{Me, } R^{4} = \text{Et}^{2} \\ 21e \ R^{1} = R^{2} = R^{3} = \text{Me, } R^{4} = \text{Et}^{2} \\ 21e \ R^{1} = R^{2} = R^{3} = \text{Me, } R^{4} = \text{Et}^{2} \\ 21e \ R^{1} = R^{2} = R^{3} = \text{Me, } R^{4} = \text{Et}^{2} \\ 21e \ R^{1} = R^{2} = R^{3} = \text{Me, } R^{4} = \text{Et}^{2} \\ 21e \ R^{1} = R^{2} = R^{3} = \text{Me, } R^{4} = \text{Et}^{2} \\ 21e \ R^{1} = R^{2} = R^{3} = \text{Me, } R^{4} = \text{Et}^{2} \\ 21e \ R^{1} = R^{2} = R^{3} = \text{Me, } R^{4} = \text{Et}^{2} \\ 21e \ R^{1} = R^{2} = R^{3} = \text{Me, } R^{4} = \text{Et}^{2} \\ 21e \ R^{1} = R^{2} = R^{3} = \text{Me, } R^{4} = \text{Et}^{2} \\ 21e \ R^{1} = R^{2} = R^{3} = \text{Me, } R^{4} = \text{Et}^{2} \\ 21e \ R^{1} = R^{2} = R^{3} = \text{Me, } R^{4} = \text{Et}^{2} \\ 21e \ R^{1} = R^{2} = R^{3} =$$

tween piperazine and phthalimide fragments, however the optimum carbon chain contains 5–6 atoms [36] (Scheme 15).

Among the studied aryl or hetepoarylpurines, 6-(2-furyl)purines are most efficient. Furthermore, the presence of a substituent at the position 9 was a crucial factor concerning the activity against *M. tuberculosis*. Compounds with small alkyl groups or with no substituent at position 9 are inactive, 9-aryl- and 9-sulphonylarylpurines exhibit a low activity against mycobacteria, however 9-benzylpurines represent good inhibitors, particularly those of them which have electron donating substituents in

the phenyl ring. However, an insignificant increase in the distance between purine and phenyl rings (CH₂CH₂Ph) resulted in the reduction of the activity. The growth of the activity is promoted also by chlorine atom at the position 2 of purine. The leading structures of this class of compounds are presented by 2-chloro-6-(2-furyl)-9-benzylpurine **24a** (MIC = 0.78 μ g/mL) and 2-chloro-6-(2-furyl)-9-(4-methoxyphenylmethyl)-9*H*-purine **24b** (MIC = 0.39 μ g/mL in contrast with as compared to the MIC value for rifampicin equal 0.25 μ g/mL). Compound **24a** exhibits low toxicity and high activity with respect to several monoresistant strains of *M*.

$$R^{1}R^{2}NH^{-} = n-\text{hexyl-NH-}, n-\text{octyl-NH-} \text{ (22a)}, \\ n-\text{dodecyl-NH-} \text{ (22b)}, morpholinyl, \\ phenylmethyl-NH-, 3-\text{chlorophenylethyl-NH-}, \\ 4-\text{methoxyphenyl-NH-}, 2-\text{hydroxyethyl-NH-} \\ n = 3-5, 7, 10 \text{ (22c)}$$

$$N - (CH_{2})_{2} - N$$

$$N - (CH_{2})_{2} - N$$

$$N = 0, 1, 2 \text{ (23a)}, 3 \text{ (23b)}$$

Scheme 15.

Scheme 14.

tuberculosis. A high antituberculosis activity with respect to *M. tuberculosis* H37Rv, low toxicity for human cells and the activity inside macrophages inherent in the compound **24b** indicate the compound is possible to use as an antituberculosis remedy [37, 38] (Scheme 16).

A rather interesting activity with respect to $M.\ tuberculosis$ H37Rv is exhibited by 9-sulphenyl /sulfonyl-6-mercaptopurines (MIC = $0.39-3.39\,\mu g/mL$). In this connection, these compounds were chosen as leading structures for obtaining much more efficient antituberculosis remedies. Compound **24c** exhibits the activity against several drug-resistant strains of $M.\ tuberculosis$ (MIC < $1\,\mu g/mL$) [39].

Amongst thio derivatives of purine, pyrimidine and pyridine tested for the activity with respect to *M. tuberculosis* strains H37Rv and H37Ra, the compounds of pyrimidine and pyridine appeared inactive. Thio compounds of purine with a substituent at the position 6 are much more active than their 6,9-substituted analogues among those 9-(ethylcarboxymethyl)-6-

$$X \xrightarrow{\text{Ar}} X \xrightarrow{\text{6}} X \xrightarrow{\text{5}} X \xrightarrow{\text{7}} X \xrightarrow{\text{8}} X \xrightarrow{\text{2}} X \xrightarrow{\text{4}} X \xrightarrow{\text{9}} X \xrightarrow{\text{8}} X \xrightarrow{\text{8}} X \xrightarrow{\text{8}} X \xrightarrow{\text{9}} X \xrightarrow{\text{8}} X \xrightarrow{\text{9}} X \xrightarrow$$

 $\begin{array}{lll} X &= H, & Cl; \\ Ar &= 2\text{-thienyl}, & 2\text{-furyl}, & Ph; \\ R &= H, & CH_3, & CH_2CH=CH_2, \\ & CH_2Ph, & CH_2\text{-cyclohexyl}, \\ & CH_2CH_2Ph, & THP; \end{array}$

$$X^{\frac{6}{2}}$$
 $X^{\frac{3}{2}}$
 $X^{\frac{3}{4}}$
 $X^{\frac{3}{4}}$

 $\begin{array}{l} X = H \\ R^2 = H, \quad OCH_3, \quad Cl, \quad F; \quad R^3 = H, \quad CH_3, \quad OCH_3, \quad Cl; \\ R^4 = H, \quad CH_3, \quad C(CH_3)_3, \quad CF_3, \quad NH_2, \quad NHCOCH_3, \\ \quad N(CH_3)_2, \quad NO_2; \quad SCH_3, \quad SO_2CH_3; \quad OH, \quad OCH_3, \\ \quad OCH_2CH_3, \quad OCH_2Ph, \quad OCOCH_3; \quad Cl, \quad F; \\ R^5 = H, \quad OCH_3; \quad R^6 = H, \quad Cl, \quad F; \end{array}$

 $X = H, CCH_3, R = H, CH, F,$ X = Cl $R^2 = H, CCH_3; R^3 = H, CH_3, CCH_3, Cl;$ $R^4 = H, CH_3, CCH_3, Cl, F;$ $R^5 = R^6 = H;$ $24a R^2 = R^3 = R^4 = R^5 = R^6 = H;$

24b $R^2 = R^3 = H$, $R^4 = OCH_3$, $R^5 = R^6 = H$

Scheme 16.

(decylthio)-9H-purine **24d** (MIC = 1.56 µg/mL) and 9-(ethylcarboxymethyl)-6-(dodecylthio)-9H-purine **24e** (MIC = 0.78 µg/mL) are most remarkable. It is obvious that substitution at the position 9 of the purine ring a intensifies the antituberculosis activity of these compounds [40].

A good inhibiting ability with respect to M. tuberculosis H37Rv is exhibited pyrazolo[3,4-d]pyrimidines **24g-i,l** and **24f,j,k,m** (MIC = 1.2 and 3.25 μ g/mL, respectively) [41] (Scheme 17).

The authors of [43] synthesized and tested for antituberculosis activity (against M. tuberculosis H37Rv) different 7-chloroquinolone derivatives, designed via joining together the two pharmacophore fragments of into one molecule antimalarial preparations and ethambutol. The important property of these compounds is presented by the absence of cytotoxicity at the concentration efficient for the inhibition of *M. tuberculosis*. The most active among the compounds studied were 7-chloro-4-diaminoquinoline derivatives 25a-c, wherein an increase in the length of the alkyl chain between nitrogen atoms results in improving the activity (n = 6, 8, 10, MIC = 25.00, 6.25,3.12 µg/mL, respectively).

A similar approach was used by the authors of [44]. The derivatives designed on the base of isoniazid (MIC = $0.2 \,\mu\text{g/mL}$) and *trans*-cinnamic or benzoic acids exhibit antituberculosis activity being herewith not cytotoxic. All the compounds of benzoic acid are sufficiently more active than their own cinnamic analogues **25d–g** (MIC = $3.12 \,\mu\text{g/mL}$).

The presence of the cinnamic acid fragment is very important for the antituberculosis activity of its linear thioesters and amides containing substituents at the position 4 of the benzene ring **25h-k** (MIC = 0.6, 0.6, 0.1, $1 \mu g/mL$ with respect to *M. tuberculosis* H37Rv) [44] (Scheme 18).

Among 20 quinoline derivatives containing different substituents (triazole, urea, thiourea), the three compounds **25l-n** were chosen with functional groups required for the presence of the antituberculosis to activity (MIC = 3.125, 6.25, $3.125 \,\mu\text{g/mL}$, respectively, against *M. tuberculosis* H37Rv) [45] (Scheme 19).

The structural optimization of 4-(adamantane-1-yl)-2-quinolinecarbohydrazide and 4-(adamantane-1-yl)-2-quinolinecarboxamide studied earlier, allowed researchers to reveal compounds 250-t (MIC = $3.125 \, \mu g/mL$ against a

$$\begin{array}{c} S \\ HN_{1} \\ \frac{1}{3} \\ \frac{1}{3} \\ \frac{1}{4} \\ N \\ O = S = O \\ R \\ R \\ R = 4-NO_{2}C_{6}H_{4}, \\ R = N(CH_{3})_{2}, X-C_{6}H_{4} \\ X = 4-CH_{3}, 4-NH_{2}, 4-NHA_{C}, \\ 4-Cl, 4-Br, 4-F, 3-NH_{2}, 2-NO_{2} \\ \end{array} \begin{array}{c} R = 4-NO_{2}C_{6}H_{4}, \\ 2-NO_{2}C_{6}H_{4} \\ (24c) \\ X = 4-CH_{3}, 4-NH_{2}, 4-NHA_{C}, \\ 4-Cl, 4-Br, 4-F, 3-NH_{2}, 2-NO_{2} \\ \end{array} \begin{array}{c} X = H, OH \\ R = C_{10}H_{21}, C_{12}H_{25}, 4-ClC_{6}H_{4}, \\ 3-MeC_{6}H_{4}, 4-MeC_{6}H_{4}, 2-naphthyl, \\ CH_{2}C_{6}H_{3}-2-Me-3-NO_{2}, CH_{2}C_{6}H_{4}-2-F \\ 24d \\ X = H, \\ R = C_{10}H_{21}; \\ 24e \\ X = H, \\ R = C_{10}H_{21}; \\ 24e \\ X = H, \\ R = C_{10}H_{21}; \\ 24e \\ X = H, \\ R = C_{12}H_{25} \\ \end{array}$$

Scheme 18.

susceptible strain of M.tuberculosis H37Rv) and compound **25u** (MIC = 1.00 μ g/mL with respect to a drug-susceptible strain of M.tuberculosis H37Rv and 3.125 μ g/ mL with respect to an isoniazid-resistant strain of M.tuberculosis H37Rv) [46] (Scheme 20).

The analogue of new-generation fluoroquinolones such as oxoquinoline **26a** (MIC = $0.2 \,\mu\text{g/mL}$) is sevenfold more active against *M. tuberculosis* with MDRTB than isoniazid (MIC = $1.56 \,\mu\text{g/mL}$) [47]. The antituberculosis activity

of novel lipophilic fluoroquinolone derivatives obtained via the substitution of the 1,2-diamine fragment by N-alkylated 1,2-ethanediamine or 1,3-propanediamine depends on the lengths and brunching level of the alkyl chain. The ideal carbon chain contains 10 atoms, as it is for compounds **26b** (MIC = $0.62 \, \mu g/mL$ with respect to M. tuberculosis H37Rv) and **26c** (MIC = $0.3 \, \mu g/mL$). All the 1.3-propanediamine derivatives are much more active comparing to 1,2-ethanediamine derivatives with similar alkyl chains [48].

$$\mathbb{R}^3$$
 \mathbb{N}
 \mathbb{N}

251 $R^3 = 3$ -OMe-Ph, X = O**25m** $R^3 = 3$ -NO₂-Ph, X = O

$$\begin{array}{c} N = N \\ Ph \\ N \\ N \\ O \\ R^2 \end{array}$$
 25n R² = imidazolyl

Scheme 19.

250
$$R^1 = \bigvee_{H} R^2 = CO_2C(CH_3)_{2}$$

25p
$$R^1 = \bigvee_{N \searrow NH} R^2 = CO_2C(CH_3)_{\xi}$$

 $\begin{aligned} \textbf{25q} \ \ R^1 &= (CH_2)_3 NHC (= NH) NH_2 \\ R^2 &= CO_2 C (CH_3)_3 \end{aligned}$

Adamantane
$$\begin{array}{c} & & \\$$

25r
$$R^1 = N NH R^2 = OCH_3$$

 $\begin{aligned} \mathbf{25s} \ \ R^1 &= (CH_2)_3 NHC (= NH) NH_2, \ \ R_2 &= OCH_3 \\ \mathbf{25t} \ \ R^1 &= (CH_2)_3 NHC (= NH) NH_2, \ \ R_2 &= NHNH_2 \end{aligned}$

Scheme 20.

The class of $6-\infty$ -6,9-dihydro-3H-[1,2,3]triazole[4,5-h]quinoline-7-carboxylic acids and their esters is structurally close to fluoroquinolones, however they contain the triazole ring, which could affect on the lipophilic properties or the activity of the molecule as a whole. Indeed, some of these compounds exhibit the antituberculosis activity that correlates with the length and position of the substituent in the triazole ring, as well as depends on the type of the substituent at the quinolone nitrogen atom. A methyl substituent at the N-3 atom of the triazole ring appeared to exhibit optimum properties. Its substitution by the ethyl group sharply reduces the activity (MIC₉₀ from 1.6 to >32 µg/mL), however when transferring the methyl groups from N-3 atom towards N-2 or N-1 atom the activity disappears. The presence of the alkyl substituent at N-9 atom appeared to be much more preferred for high antituberculosis activity than the presence of propenyl or benzyl groups, whereas the phenylethyl group is tolerant one. The most efficient derivative possessing a high antituberculosis activity associated with the absence of cytotoxicity, appeared to be presented by compound 26d $(MIC_{90} = 0.5 \,\mu g/mL \text{ with respect to } 11 \text{ clinical})$ isolates of M. tuberculosis and human infectious macrophages (J774-A1) [49].

Novel derivatives of fluoroquinolone carboxylic acids **26e-g** exhibit a pronounced antituberculosis activity, which determines their prospectivity in the further searching for antituberculosis remedies. Fluoroquinolones **26e** (MIC = $0.2-1.6~\mu g/mL$) exhibit a much more high antituberculosis activity than perfloxacin (MIC = $4~\mu g/mL$).

In the series of thiadiazinoquinolones **26f** (MIC = $0.2-1.6 \,\mu g/mL$), the activity decreases to a considerable extent when substituting $R^3 = H$ by fluorine atom. Among oxadiazinoquinolones **26g** (MIC = $0.2-0.6 \,\mu g/mL$), the compound having nitrophenyl remainder as R^1 has demonstrated somewhat greater activity as compared to pyridine-4-yl-substituted derivatives [50] (Scheme 21).

A highly efficient inhibitor of indole-3-glycerol phosphate synthase (IGPS) of *M. tuberculosis* is presented by compound **27** (ATB107) that is rather active both with respect to the laboratory strains of *M. tuberculosis* H37Rv, *M. tuberculosis* H37Ra and *M. bovis* BCG (MIC

0.1 μ g/mL), and with respect to clinical isolates such as drug-susceptible strains (50 strains, MIC = 0.1–1.0 μ g/mL) and strains with MDRTB (80 strains, MIC = 1.0 μ g/mL). The further studies concerning the toxicity, pharmacology and activity *in vivo* in model animals are under planning [51] (Scheme 22).

Oxygen-containing heterocycles

As a base structure for synthesizing the second nitrofuranylamide generation, 3,4-dimethoxybenzylamide of 5-nitrofuran-2-carboxylic acid **28a** found earlier exhibiting a significant antituberculosis activity (MIC = $0.2 \, \mu g/mL$), but having a low solubility, was chosen. The authors put the problem to improve the solubility

COOEt

Scheme 22.

and bioavailability of novel compounds due to the introducing hydrophilic cyclic fragment C to the benzyl or phenyl ring B. It appeared that substituted benzyl compounds (MIC = 0.0125– $3.13 \,\mu\text{g/mL}$) exhibit a more high antituberculosis activity, than substituted phenyl compounds (MIC = 0.4– $12.5 \,\mu\text{g/mL}$). In both cases,

$$\begin{array}{c|c} F & O & O \\ \hline R & N & OH \end{array}$$

 $\mathbf{26a-c} \ R = Me, \ NH(CH_2)_2NHC_{10}H_{21}, \ NH(CH_2)_3NHC_{10}H_{21}$

 $\begin{array}{c|cccc}
R^4 & O & O \\
\hline
R & & & & \\
R^3 & & & & \\
R^2 & & & & \\
\hline
\mathbf{26e} & & & & \\
\end{array}$

 $R = NHNH_2$, OH, OEt

 $\mathbf{R}^1 = \mathbf{Et}, \ \mathbf{NHC(O)CH_2CN}, \ 2\text{-}\mathbf{Cl}\text{-}4\text{-}\mathbf{CF_3}\text{-}\mathbf{Ph}$

 $R^2 = H$, F $R^3 = morpholine-4-yl$, 2,6-morpholine-4-yl, pyrrolidine-1-yl, 4-methylpiperidine-1-yl

26d

 $R^4 = H$, F, pyrrolidine-1-yl

R = OH

 R^1 = pyrrolidine-1-yl

 $R^2 = F$, 4-ethoxycarbonyl, pyrrolidine-1-yl, morpholine-4-yl, pi perazine-1-yl

R³ = H, F, 4-ethoxycarbonyl, morphline-4-yl, morpholine-4-yl, piperazine-1-yl

26g

R = OH, OEt

 $R^1 = 3$ -nitrophenyl, pyrrolidine-1-yl

 $R^2 = F$, pyrrolidine-1-yl

 $R^3 = H, F$

Scheme 21.

substitution at para position caused the antituberculosis activity to be improved. The compound from the benzyl series are extremely efficient, in particular compound **28b** with a $parabenzylpi perazine substituent (MIC = <math>0.0125 \, \mu g/mL$ against $M. \, tuberculosis \, H37Rv)$ and compound **28c** similar to the mentioned one with meta-fluorine atom in the benzene ring (MIC = $0.025 \, \mu g/mL$) [52] (Scheme 23).

The authors of [53] synthesized potential antituberculosis compounds having in the structure the dibenzofuran skeleton and dimethvlpyrane ring condensed together with it. Compound 29a (MIC = $5 \mu g/mL$) and its reduced analogue 29b (MIC = $1-5 \mu g/mL$) actively inhibit the growth of different strains of M. tuberculosis alongside with isoniazid. Furthermore, compounds obtained exhibit a low cytotoxicity level. It was expected that the double bond in the pyrane cycle would allow additional potentialities for the modification. However, the function alization of these compounds aimed at improving the solubility in biocompatible solvents, namely the dihydroxylation of the dimethylpyrane ring and the further conversion into esters resulted in a complete loss of the activity. As the result of studies concerning the optimization of structure–antituberculosis activity relations, compounds **29c** and **29d** were found inhibiting well M. tuberculosis (MIC = 2.5 and $3 \,\mu g/mL$, respectively) and M. smegmatis (MIC = 3 and $6.2 \,\mu g/mL$, respectively) [54], as well as furo[3,2-f]-chromanes **29e** (MIC₉₅ = 0.6–5 $\,\mu g/mL$ for M. bovis BCG and M. tuberculosis H37Rv) [55]. Unfortunately, the majority of compounds **29e** at these concentration values are cytotoxic (Scheme 24).

The following group of compounds is presented by synthetic derivatives of saccharides.

Among the compounds those represent the C-phosphonate analogues of decaprenolphosphoarabinose being key intermediates in the biosynthesis of mycobacterial arabinogalactan and lipoarabinomannan, only compound 30 (MIC = $3.13\,\mu\text{g/mL}$) with a hexadecyl substituent exhibits the required activity with respect to M. tuberculosis H37Rv ATCC 27294. This fact clearly indicates that the length of the alkyl chain determines the antituberculosis activity of this type of compounds [56]. A novel class of compounds such as derivatives of 2,3-dideoxy-hex-2-en-pyronazide with alkyl and arylalkyl substituents at C-3 atom of hexen-pyranoside was studied in vitro aimed at com-

O2N OMe O2N Phenylamides
$$O_{2}N$$
 $O_{2}N$ $O_{$

29a 29b

29c
$$R^1 = 4$$
-pyridyl, $R^2 = H$
29d $R^1 = 4$ -pyridyl, $R^2 = Me$
29e $R^1 = 4$ -pyridyl, $R^2 = Me$
3-cH₂-pyridyl, $R^2 = Me$
3-pyridyl-(CH₂)₂-,

Me₂N(H₂C)₃HN $R^2 = Me$

Me₂N(H₂C)₃HN $R^2 = Me$

Me₂N(H₂C)₃HN $R^2 = Me$

Scheme 24.

plete inhibiting the growth of M. tuberculosis H37Rv. For the further studying, compound **31** (MIC = $3.12 \, \mu g/mL$) was chosen which is characterized by a maximal activity combined with the absence of cytotoxicity [57].

The antituberculosis activity was studied for the series of N- and C-alkylated amino alcohols and their galactopyranozyl derivatives 32a-i. It was demonstrated that the activity free amino alcohols depends on the length of the alkyl chain: the best results were obtained for compounds 32a and 32b. Herewith the C-alkylated compound 32b (MIC = $3.12 \mu g/mL$ with respect to M. tuberculosis H37Rv) is two times more active than its N-alkylated analogue 32a (MIC = $6.25 \,\mu\text{g/mL}$). Compounds with a longer alkyl chain appeared two times less active comparing to compound 32a, but more active than compounds with a shortened alkyl chain. A similar effect was observed also for galactopyranozyl derivatives of these amino alcohols, but in this case N-alkylated galactoso derivative 32c (MIC = $3.12 \,\mu g/mL$) is four times more active than C-alkylated analogue 32d (MIC = $12.5 \mu g/$ mL). As a whole, the glycosylation of N-alkylated amino alcohols enhanced their activity indicating that the carbohydrate fragment is important for antituberculosis activity of these compounds. One more important factor determining the activity, is presented by the linearity of the hydrocarbon chain, since the compound those have a brunched alkyl chain do not exhibit antituberculosis activity at the concentration under testing [58].

The authors of [59] noted that the antituberculosis activity of glycosylated amino alcohols 32e having different length alkyl chain depends on the nature of the hydroxyaminoalkyl chain. So, compounds with simple linear aminoalkyl chain more active than their analogues with furfuryl, benzyl or cyclic amine fragment. Compound 32f (MIC = $1.56 \mu g/$ mL) chosen as a leader in order to search for much more efficient analogues surpasses in the action the existing antituberculosis preparations such as amino alcohol ethambutol with respect to M. tuberculosis H37Rv (MIC = $3.25 \mu g/mL$) and with respect to five clinical isolates with MDRTB at the concentrations of 50 µg/mL wherewith the existing antituberculosis preparations are inefficient. Such an activity requires for 12-membered carbon chain and the presence of two galactopyranozyl units.

A novel series of amino sugar derivatives was tested for antituberculosis activity against M. tuberculosis H37Ra and H37Rv. It was established that compound 32g (MIC = $3.12 \, \mu g/$ mL) with N-dodecyl and 3-O-benzyl substituents exhibits the best antituberculosis activity amongst the compounds this series. Galactopyranosylated amino alcohol 32h (MIC = $3.125 \, \mu g/$ mL) with a hexadecyl substituent appeared the most active against the strain of M. tuberculosis H37Rv, whereas compound 32i (MIC = $3.12 \, \mu g/$ mL) with oleyl fragment was most active with respect to M. tuberculosis

H37Ra. For obtaining compounds with lower values of MIC, further optimization is necessary [60] (Scheme 25).

Sulphur-containing heterocycles

A novel type of compounds with antimycobacterial activity is presented by the derivatives of triarylmethane 33a-f (MIC = $3.12 \,\mu g/mL$ with respect to M. tuberculosis H37Rv), wherein one aryl group is substituted by the thiophene fragment, that exhibits aromatic properties, too [61].

 $\begin{array}{lll} R = (CH_2)_6 CH_3, & (CH_2)_7 CH_3, & (CH_2)_8 CH_3, \\ & (CH_2)_9 CH_3, & (CH_2)_{11} CH_3, & (CH_2)_{15} CH_3 & \textbf{(30)} \end{array}$

 $CH_3(CH_2)_nNHCH_2CH_2OH$ n = 7, 9, 11 (32a), 13, 15

32c

OH
NHR¹

OR

R = Me, CH₂Ph
NHR¹ = HN-cyclopropyl, HN

32d

 $\begin{array}{l} {\rm H-Me,\ CH_2^{1}\ II} \\ {\rm NHR^{1}=HN-cyclopropyl,\ HN-cyclohexyl,} \\ {\rm HN(CH_2)_6CH_3,\ HN(CH_2)_{11}CH_3,} \\ {\rm HN(CH_2)_{15}CH_3,\ oleylamine} \\ {\rm 32g\ R=(CH_2)Ph,\ NHR^{1}=HN(CH_2)_{11}CH_3} \end{array}$

 $\begin{array}{c} CH_{3}(CH_{2})_{9}CH(OH)CH_{2}NH_{2}\\ \\ \textbf{32b} \end{array}$

 $\begin{array}{llll} \textbf{32e} & \text{NR}^1\text{R}^2 = \text{PhCH}_2\text{NH-, furfuryl NH-,} \\ & \text{CH}_3(\text{CH}_2)_{11}\text{NH-, H}_3(\text{CH}_2)_{15}\text{NH-,} \\ & \text{CH}_3(\text{CH}_2)_7\text{CH=CH}(\text{CH}_2)_8\text{NH-,} \\ & \text{1-pyrrolidinyl, 1-piperidinyl,} \\ & \text{1-morpholinyl,} \\ & \text{4-(6-chloro-2-pyridyl)piperazinyl} \end{array}$

$$\begin{split} \mathrm{NHR}^1 &= \mathrm{HN-cyclopropyl},\\ \mathrm{HN-cyclohexyl},\\ \mathrm{HN}(\mathrm{CH}_2)_6\mathrm{CH}_3,\\ \mathrm{HN}(\mathrm{CH}_2)_{11}\mathrm{CH}_3,\\ \mathrm{HN}(\mathrm{CH}_2)_{15}\mathrm{CH}_3 \ \textbf{(32h)},\\ \mathrm{olevlamine} \ \textbf{(32i)} \end{split}$$

33a-f R^1 = 4-OMe, 3-OMe, 4-SMe, 4-Cl R^2 = OH, NEt₂, heptamethyleneimine

It is known for long that phenothiazines exhibit the antituberculosis activity, but their using is limited by psychotropic effects (medicinal preparations chlorpromazine, tirioridazine, trifluoperazine), which could be caused by binding these compounds with dopamine and serotonin receptors. In connection with these circumstances, the synthesis of phenothiazine analogues was realized for the purpose of clarifying of the types of their modification those result in increasing the antituberculosis activity and action selectivity. All the existing phenothiazine preparations possess different lateral chains in the structure, at the position 10 of phenothiazine ring, as well as substituents in the aromatic ring. Furthermore, for some classes of the compounds, the events are known connected with increasing the biological activity, when minimal active units are bound between each other with the formation of dimers or bis-compounds.

For a series of compounds with a fixed by methylpi peridine lateral chain, the most active compounds with respect to M. tuberculosis H37Rv are presented by the compounds with phenyl substituents 34a and 34b (MIC = 4.5 and 2.1 µg/mL, respectively) those exhibit a weak tendency towards binding with serotonin receptors and only moderate tendency towards binding with dopamine receptors. The most considerable tendency towards binding with these receptors is inherent in analogues 34c and **34d**, inactive with respect to *M. tuberculosis*. Among all the compounds with fixed by CF₂ substituent in the aromatic part of the molecule and varied lateral chain, one could indicate only two compounds (34e and 34f, MIC = 4.6 and 4.2 μg/mL, respectively). An increase in the steric volume of the lateral chain in the compound 34g-j (MIC = $10-20 \mu g/mL$) causes the binding level with all the subtypes of receptors to decrease. Only both bisphenothiazine compounds 34k and 34l (MIC = 2.3 and 2.0 μg/mL, respectively) simultaneously demonstrate both the most significant increase

in the antituberculosis activity, and a considerable reduction of binding with a receptor [62] (Scheme 26).

Joining together the cycle of 4H-1,2,4benzothiadiazine-1,1-dioxide with the pyridine fragment those separately exhibit antimicrobial and antituberculosis effect, allowed researchers to obtain a novel type of a structure with antituberculosis activity. Compound 35 demonstrates the best activity with respect to M. tuberculosis H37Rv ATCC 27294, drugresistant and drug-susceptible clinical isolates of M. tuberculosis (MIC = $0.5-2.0 \,\mu\text{g/mL}$), as well as a moderate activity with respect to M. avium ATCC 49601 and M. intracellulare ATCC 13950 (MIC = $2.0 \,\mu\text{g/mL}$). However, this compound has not demonstrated such an activity when in vivo testing in mice, to all appearance, due to bad bioavailability [63].

In order to optimize the structure—antituberculosis activity relationship, different alkyl esters of 2- and 3-[5-(nitroaryl)-1,3,4-thiadiazol-2-yl-thiosulfinyl- and sulfonyl]acetic and propionic acids were synthesized, in this case the type of aryl substituent (5-nitroheterocycle), the oxidation degree of sulphur atom (n=0-2), the presence of methyl and methylene groups at α -carbon atom of the estereal fragment, the structure ester groups were varied. As a result, compound **36** was found with a high antituberculosis activity against *M. tuberculosis* H37Rv (MIC = 1.56 µg/mL), which suits for the further *in vitro* and *in vivo* estimation [64] (Scheme 27).

The activity with respect to *M. tuberculo*sis H37Rv and the strains of M. tuberculosis with MDRTB was studied for 22 novel 2-aryl-3,4-dihydro-2*H*-thieno-[3,2-*b*]indoles. Among the compounds of 13 indoles tested, compounds 37a-n inhibit the growth of M. tuberculosis much more efficiently than the applied medicinal preparations. So, compounds 37e,f,h,l,n (MIC = 0.78, 0.78, 0.78, 0.4, 0.78 μg/mL, respectively, with respect to M. tuberculosis H37Rv) are more active, than ethambutol (MIC = $1.56 \,\mu g/mL$) and pyrazinamide (MIC = $6.25 \,\mu g/mL$), but compounds 37a,b,h,i and 37e,f,l,o (MIC with respect to M. tuberculosis with MDRTB being equal to 0.78 and 0.4 µg/mL, respectively) are more efficient than isoniazid (MIC = $1.56 \mu g/mL$) and pyrazinamide (MIC = $3.13 \,\mu g/mL$). The results obtained demonstrate that the increase in the

34a-d R = 3-Ph, phenoxazine, 2-Ph, 3-morpholino

$$rac{1}{1}$$

$$\mathbf{34e}\text{-}\mathbf{g} \ \mathbf{R} = \begin{array}{c} \mathbf{NH} \\ \mathbf{OH} \end{array}, \quad \begin{array}{c} \mathbf{NH} \\ \mathbf{NH} \\ \mathbf{N} \end{array}$$

$$\begin{array}{c|c} CF_3 & CF_3 \\ \hline \\ N & N \\ \hline \\ S & \\ \hline \\ 34k \\ \end{array}$$

Scheme 26.

$$\begin{array}{c|c}
O & O \\
S & O \\
N & M \\
Me & H \\
35
\end{array}$$

$$Ar \xrightarrow{N-N}_{S} COOR$$

36 Ar = 5-nitro-2-thienyl; R = n-Pr

Scheme 27.

 $37a \text{ Ar} = 2,4\text{-Cl}_2\text{C}_6\text{H}_3; X = \text{H}$

 $\begin{array}{l} \textbf{37b} \ Ar = 3\text{-NO}_2C_6H_4; \ X = H \\ \textbf{37c} \ Ar = 2\text{-Br}C_6H_4; \ X = H \\ \textbf{37d} \ Ar = 3\text{-FC}_6H_4; \ X = H \\ \textbf{37d} \ Ar = 4\text{-Pr}^iC_6H_4; \ X = H \\ \textbf{37f} \ Ar = 2\text{,}4\text{-Cl}_2C_6H_3; \ X = 7\text{-Cl} \\ \textbf{37g} \ Ar = 2\text{-Cl}C_6H_4; \ X = 7\text{-Cl} \\ \textbf{37h} \ Ar = 4\text{-Pr}^iC_6H_4; \ X = 7\text{-Cl}$

37i Ar = 4-FC₆H₄; X = 7-Cl 37j Ar = 4-ClC₆H₄; X = 7-F

37k Ar = 4-MeC₆H₄; X = 7-F 37l Ar = 2,4-Cl₂C₆H₃; X = 7-F

37m Ar = $2 - \text{ClC}_6 \text{H}_4$; X = 7 - F

37n Ar = $4 - Pr^{i}C_{6}H_{4}$; X = 7-F

Scheme 28.

activity is promoted by the presence of a halogen atom in the ring, in this case fluorinated derivatives are more active, than chlorine-containing compounds [65] (Scheme 28).

Other compounds

A very good antituberculosis activity comparable with the activity of ethambutol (MIC = $3.12 \,\mu\text{g/mL}$) is exhibited by α, ω -diaminoalkanes $\text{H}_2\text{N}(\text{CH}_2)_n\text{NH}_2$ 38a-c (n=9-12, MIC = $2.5-3.12 \,\mu\text{g/mL}$ with respect to M. tuberculosis H37Rv). Moreover, these compounds are not cytotoxic at the concentration efficient for the inhibition of M. tuberculosis. It was established that for the activity to exhibit, both certain length of alkyl chain, and two amino groups in free state are required [66].

The derivatives of dihydrocsphingosine 39a–c (MIC = $1.25 \,\mu\text{g/mL}$) exhibit a high antimycobacterial activity with respect to strain $M.\ tu$ -berculosis H37Rv and clinical isolates, resistant against five medicinal antituberculosis preparations belonging to the first line, as well as 11 different strains with MDRTB. One should especially note compound 39b which inhibits the

growth of strain 332 resistant with respect to all the medicinal antituberculosis preparations belonging to the first line, with a much lower MIC value (0.78 μ g/mL) [67].

$$\begin{array}{c} R^1 \\ n\text{-}C_{14}H_{29} \\ \\ \textbf{39a} \ R^1 = \text{OBn}, \ R^2 = \text{NH}_2 \\ \textbf{39b} \ R^1 = \text{OBn}, \ R^2 = \text{NEt}_2 \\ \textbf{39c} \ R^1 = \text{NEt}_2, \ R^2 = \text{NH}_2 \end{array}$$

It is known that 5-hexyl-2-phenoxyphenol $40a~(\mathrm{MIC}_{90}=(2.1\pm0.9)~\mu\mathrm{g/mL})$ inhibits well the growth of M.~tuberculosis strains both drugsusceptible, and drug-resistant with respect to isoniazid, however it has a considerable disadvantage: pronounced lipophilic properties of this compound cause the bioavailability to be reduced. The analogue of this compound $40b-d~(\mathrm{MIC}_{90}=3.13~\mu\mathrm{g/mL})$ obtained by modification of the ring B, the level of lipophilic properties was succeeded to reduce with the level of antituberculosis activity being conserved [68].

Ortho-, meta- and para-substituted diphenyl esters **40e-n** (MIC = 1 μ g/mL) are also highly active with respect to *M. tuberculosis* H37Rv [69].

Optimizing the structure of triclosan (MIC = $40 \,\mu g/mL$ for M. tuberculosis H37Rv) was performed. The compounds obtained exhibit an improved antituberculosis activity, too, herewith the inhibitors those have hydrophobic substituents such as alky groups, appeared much more efficient. The derivatives of triclosan **40o,p** (MIC = $4.7 \,\mu g/mL$) demonstrate a good activity only with respect to clinical isolate 5071 (resistant against isoniazid and streptomycin), whereas compound **40q** is efficiently both against widespread M. tuberculosis H37Rv, and against clinical isolate 5071 with similar MIC values ($4.7 \,\mu g/mL$) [70].

Very important parameters for a good antituberculosis activity of salicylanilides **40r,s** (MIC = $3.13 \, \mu \text{g/mL}$, for isoniazid MIC = $0.025 - 0.2 \, \mu \text{g/mL}$) with respect to *M. tuberculosis* H37Rv (ATCC 27294) are presented by the substituent R¹ (CF₃ group and atom Cl at certain of the aromatic ring), small size and the stereochemistry of substituent R² (*S*-methyl group). In order to determine an exact difference in the antituberculosis activity between individu-

al R/S enantiomers, additional studies are required [71] (Scheme 29).

Active inhibitors of growing M. tuberculosis H37Rv such as 41a-d (MIC = $0.79-3.57 \mu g/mL$, for rifampicin MIC = $0.125 \mu g/mL$) were revealed among 5-methyl/trifluoromethoxy-1H-indole-2,3-dione-3-thiosemicarbazones [72].

A high efficiency with respect to M. tuberculosis H37Rv and clinical isolates of M. tuberculosis with MDRTB, selectivity of actions and low cytotoxicity are inherent in 4-(5-cyclobutyloxazole-2-yl)thiosemicarbazones. So, with respect to M. tuberculosis H37Rv, the action of compound 42 equivalent to isoniazid (MIC = 0.05 µg/mL for both compounds), whereas eight compounds are much more efficient (MIC = $0.05-0.78 \,\mu g/mL$) than ethambutol (MIC = 1.56 µg/mL). Eight compounds inhibit the growth of M. tuberculosis with MDRTB more efficiently (MIC = $0.05-0.78 \mu g/mL$) than isoniazid (MIC = $1.56 \,\mu g/mL$), nine compounds inhibit this process more efficiently, than rifampicin (MIC = $3.12 \,\mu g/mL$), however all the compounds studied are much more efficient, than ethambutol (MIC = $25 \mu g/mL$). The antituberculosis activity the compounds under investigation depends on the presence and nature of substituents R and R^1 . When $R = R^1 = H$ the absence of inhibition mycobacteria growth was observed. Electron donating substituents R¹(NO₂, Cl) cause the activity to increase. Substituent R influences the activity according the following order: phenyl > methyl > H [73] (Scheme 30).

NATURALLY OCCURRING COMPOUNDS WITH ANTIMYCOBACTERIAL ACTIVITY

Alkines and heterocyclic compounds

The metabolite of several strains of an endophytic fungus belonging to genus *Phomopsis*, 3-nitropropionic acid **43**, actively inhibits the growth of *M. tuberculosis* H37Ra (MIC = $0.4\,\mu g/mL$). Though the strong neurotoxicity of this compound prevents it from using as a medicinal preparation, one could apply it as a useful model for the synthesis of a novel inhibitor of isocitrate lyase, an enzyme required for the catabolism of fatty acids and

Scheme 29.

the virulence of M. tuberculosis [74]. From a hexane extract of all the parts of plant Humulus lupulus, linoleic acid 44 was isolated inhibiting the growth of M. phlei (MIC = 2 μ g/mL) [75]. The polyacetylene compounds are exemplified by falcarindiol 45, isolated from Anethum graveolens ((MIC = $2-4 \mu$ g/mL when testing on the group of fast-growing mycobacteria (M. fortuitum ATCC 6841, M. smegmatis ATCC 14468, M. phlei ATCC 11758, M. aurum Pasteur Institute 104482 and M. abscessus ATCC 19977; for ethambutol MIC = $0.5-4 \mu$ g/mL) [76]. However, the cytotoxicity of this class of polyacetylene compounds could restrict the interest in their biological activity [77].

Compounds **46** and **47**, synthetic analogues of natural antibiotic thiolactomicin, inhibit the growth of M. tuberculosis with the value of $MIC = 1-16 \,\mu g/mL$ including strains with drug resistance, too [78]. The study on the components of plant $Cinnamomum\ kotoense$ resulted in the isolation of a number of compounds among those the most pronounced antituberculosis activity is exhibited by lincomolide B **48** ($MIC = 2.8 \,\mu g/mL$) [79]. Micromolide **49**, γ -lacton derivative of oleic acid was isolated from the stalk rind of $Micromelum\ hirsutum$, it is characterized by $MIC = 1.5 \,\mu g/mL$ with respect to M. tuberculosis (H37Rv). Further testing the activity with respect to the cells of mice J774,

Scheme 30.

HOOC
$$NO_2$$
43

44

OH

 C_7H_{15}

OH

 C_8H_{17}

A9

 C_8H_{17}

A9

Scheme 31.

infected with a much more virulent strain M. tuberculosis Erdman has demonstrated that value of MIC is equal to $5.6~\mu g/mL$ [80]. The 2-substituted furans $\bf 50$ and $\bf 51$ isolated from the root of Polyalthia evecta exhibit the activity against M. tuberculosis (MIC = 3.1 and $6.25~\mu g/mL$, respectively) [81]. Natural compound pamamycine-607 synthesized $\bf 52$ inhibits the growth of M. bovis BCG, M. smegmatis and M. tuberculosis (MIC = $0.5-4.7~\mu g/mL$). It was established that there is the absence of cross resistance against isoniazid and rifampicin [82] (Scheme 31).

Phenols and quinones

Phenylpropanoids 53 and 54, the metabolites of Pimpinella sp., inhibit the growth of a number of mycobacteria, including M. intracellulare, M. smegmatis, M. aurum and M. phlei (MIC 1.25-10 μg/mL) [83]. Engelhardione 54 is very active with respect to M. tuberculosis H37Rv (MIC $0.2\,\mu g/mL$) [84]. As antituberculosis components, from Engelhardia roxburghiana, (-)-4-hydroxy-1-tetralone **55** $(MIC = 4.0 \,\mu g/mL)$, 3-methoxyjuglone **56a** (MIC = $0.2 \,\mu g/mL$) and 3-methoxycarbonyl-1,5-dihydroxyanthraquinone 57 (MIC = $3.125 \,\mu g/mL$) were isolated [84]. It is known that the level of intracellular and extracellular M. tuberculosis inhibition by 7-methyljuglone **56b** (MIC = $0.5 \mu g/mL$) isolated from plant Euclea natalensis, is comparable with streptomycin and ethambutol (MIC = 1 and 2 μ g/mL, respectively). Its derivatives such as 5-hydroxy-, 5-alkoxy- and 5-acetoxy-8-substituted naphthoquinones are much more active (MIC ranging from 2.5 up to and more than 20 μ g/mL) and, moreover, exhibit a low antituberculosis selectivity, which, to all appearance, could be caused by their non-specific activity with respect to different disulphide reductases, found in the cells of mammals.

Optimization is required concerning the specificity of these compounds with respect to micothiol disulphide reductase, of several biological targets for antituberculosis activity of naphthoquinones with such type of structure [85]. Maritime metabolites pseudopyronines A and B **58a,b** (MIC = $0.78-3.125 \,\mu g/mL$) well inhibit the growth of M. tuberculosis H37Rv [86]. The component of Piper sanctum active with respect to M. tuberculosis H37Rv is presented by pyrone **59** (MIC = $4 \mu g/mL$) [87]. Ferulenol 60a isolated from the Sardinian giant fennel Ferula communis is efficient against M. smegmatis (MIC = $0.5 \mu g/mL$) as well as with respect to M. fortuitum, M. phlei and M. aurum (MIC = $2 \mu g/mL$). From the same plant, its an alogues 60b-d were isolated, among those the compound 60b with the benzyloxy group remains active with respect to M. smegmatis and M. phlei as well as against M. fortuitum and M. aurum to a lesser extent, however the activity of compounds 60c,d with hydroxy and

acetoxy groups is reduced to a considerable extent [88]. Ostruthin **60e**, the etabolite of *Peucedanum ostruthin* Koch, also inhibits the growth of *M. aurum* (MIC = $3.4 \,\mu g/$ mL) [75] (Scheme 32).

Compound 61a-h, isolated from the lichen fungus *Microsphaeropsis* sp., to a different level exhibit the activity against M. tuberculosis H37Ra (MIC = 25, 3.12, 3.126.25, 6.25, 12.5,25, 1.56-3.12, $50 \mu g/mL$, respectively), however cytotoxicity is inherent in them [89]. A dibenzofuran derivative such as usnic acid 62, a secondary metabolite of lichens, inhibits the growth of M. tuberculosis (MIC = $2.5-5 \mu g/mL$) [90]. One of xanthone dimers phomoxanthone A 63a isolated from an endophyt fungus belonging to genus Phomopsis, is very active with respect to M. tuberculosis H37Ra (MIC = $0.5 \,\mu g/mL$), whereas its deacetylated derivatives 63b is inactive. Phomoxanthone B 63c is less efficient (MIC = $6.25 \,\mu g/mL$). Both active compounds are cytotoxic [91]. Anthraquinone celastramycin B 64 isolated from an unknown Streptomyces sp., is active with respect to M. vaccae (MIC = $3.1 \,\mu g/mL$) [92]. An anti-HIV

agent (+)-calanolide A **65** was tested for the antituberculosis activity, since the existence of the same agent with anti-HIV and antituberculosis activity is very attractive, particularly taking into consideration simultaneous clinical course for the two diseases. This compound isolated from tropical tree $Calophyllum\ lanigerum$, exhibits also antituberculosis activity against $M.\ tuberculosis$ (MIC = 3.13 µg/mL) as well as against a number of drug-resistant strains (MIC = 8–16 µg/mL) [93] (Scheme 33).

Peptides

OH HC

Four cyclic peptides such as enniatin H **66a**, I **66b**, B **66c** and B4 66d, the components pathogenic fungus *Verticillium hemi pterigenum*, inhibit the growth of *M. tuberculosis* H37Ra (MIC = $3.12-6.25 \,\mu\text{g/mL}$) [94]. Syringomycin E **67** isolated from *Pseudomonas syringae* pv. *Syringae*, is active with respect to *M. smegmatis* (MIC = $1.5 \,\mu\text{g/mL}$) [95]. The metabolite of *Nocardia* sp. (ATCC 202099), the thiazole peptide nocathiacin I **68**, exhibits an ac-

$$1 \times 10^{\circ}$$
 $1 \times 10^{\circ}$
 $1 \times$

Scheme 32.

tivity with respect to M.tuberculosis ATCC 35828, M.avium A26778 and M.avium A26640 (MIC = 0.008, 0.06 and 0.25 µg/mL, respectively). Unfortunately, bad pharmacokinetics and solubility problem are inherent in this class of compounds, which could be overcome via synthesiz-

ing water-soluble analogues [96] (Scheme 34).

Alkaloids

Scheme 33.

The two compounds such as well-known antibiotic pyrrolnitrine **69a** and banegazine **69b**, isolated from zoobacteria Aristabacter necator, exhibit a synergic action against M. smegmatis (MIC > $0.5 \,\mu\text{g/mL}$ for **69b**, $0.3 \,\mu\text{g/mL}$ for **69a**, and $0.075 \,\mu\text{g/mL}$ for (**69b** + **69a**) [97]. A similar celastramycin A **70**, a dichloropyrrol metabolite of strain *Streptomyces* exhibits a broad spectrum of antimycobacterial activity (MIC = $0.05-3.1 \,\mu\text{g/mL}$ with respect to M. smegmatis, M.

aurum, M. vaccae and M. fortuitum) [98]. Quinolone alkaloids **71a-d** isolated from dried unripe fruits Evodia rutaecarpa having unsaturated aliphatic chain at the position 2 of the quinolone ring, exhibit the best antimycobacterial activity in contrast with their own analogues with a saturated lateral chain (MIC = 2-4 μg/mL with respect to M. fortuitum ATCC 6841, M. smegmatis ATCC 19429, M. phlei ATCC 19249) [75]. Bis-1-oxaquinolizidine alkaloid (¬)-araguspongine C **72** isolated from maritime sponge Xestospongia exigua, inhibits the growth of M. tuberculosis H37Rv (MIC = 1.9 μg/mL) [99]. Agelazine E **73a** and agelazine D **73b** were earlier isolated maritime sponge Agelas nakamurai.

Whereas agelazine E is inactive, its methoxy analogues 73c-e having different terpenoid lateral chains demonstrate a high activity with respect to M. tuberculosis H37Rv (MIC = 3.13, 1.56 and $3.13 \,\mu g/mL$, respectively). To all

Scheme 34.

appearance, the presence of the alkoxy group at the terminal nitrogen atom is very important factor for the antimycobacterial activity of these compounds. At the same time, only small difference was found between the activity values of agelazine D 73b and its alkoxy derivatives **73f**,**g** [100]. It is quite interesting that a simpler analogue of compounds under discussion such as 9-methyladenine 73h exhibits the MIC value, equal to $6.25 \,\mu g/mL$ [101]. Tetracyclic alkaloid cryptolepine 74a isolated from Cryptolepis sanguinolenta exhibits an activity against a number of fast-growing mycobacteria, including M. aurum (MIC = $2 \mu g/$ mL), M. phlei (MIC = $4 \mu g/mL$) and M. fortuitum (MIC = $16 \,\mu g/mL$) [102]. The metabolite of Allium neapolitanum 74c exhibits an increased activity with respect strain M. smegmatis (mc²2700) in contrast with its unsubstituted analogue cantine-6-one **74b** (MIC = 2 and $8 \mu g/mL$, respectively). It was established that the activity compound 74c against the strain M. smegmatis

67 R = NHCOCH₂CH(OH)(CH₂)₈CH₃

(mc²2700) is much higher than the activity with respect to strain M. smegmatis (ATCC 14468) (for **74c** and **74b** MIC = 16 and 8 μ g/mL, respectively) [103] (Scheme 35).

The metabolites of the Thailand pathogenic fungus Hirsutella nivea BCC 2594 hirsutellones A-D **75a-d** inhibit the growth of *M. tuberculosis* H37Ra (MIC = 0.78, 3.125, 0.78, and $0.78 \mu g/mL$, respectively). Compound 75d demonstrates moderate in vitro cytotoxicity, whereas the rest compounds are less cytotoxic [104]. Hirsutellone F 75e isolated jointly together with already known hirsutellones A, B and C from the seeds of fungus Trichoderma sp. BCC 7579, a new dimeric alkaloid, exhibits a more weak antituberculosis activity with respect to $M. tuberculosis H37Ra (MIC = 3.12 \mu g/mL) in$ contrast with hirsutellones A, B and C [105]. Already known alkaloid esteinascidin 770 76a and a new estein ascidin 786 76b isolated from Ectein ascidia thurstoni, inhibit the growth of M. tuberculosis H37Ra (MIC = 0.1 and $1.6 \mu g/mL$,

Scheme 35.

respectively) [106]. A promising antituberculosis activity is exhibited by manzamine alkaloids isolated from maritime sponges. Manzamines A **77a**, E **77c**, F **77d** and their hydroxyl derivatives, 6-hydroxymanzamine E **77e** and (+)-8-hydroxymanzamine A **77b**, exhibit an activity with respect to M. Tuberculosis H37Rv (MIC = 1.5, 3.8, 2,6, 0.4 and 0.9 µg/mL, respectively) [107]. Manadomanzamines A **78a** and B **78b** inhibit the growth of M. tuberculosis H37Rv (MIC = 1.9 and 1.5 µg/mL, respectively) [108] (Scheme 36).

Terpenoids

Compound **79** isolated from *Indigofera* longeracemosa is active with respect to M. tuberculosis (MIC = $0.38 \, \mu g/mL$) [109]. Identical MIC values with respect to M. tuberculosis H37Rv are demonstrated by diterpenoids **80** and **81** isolated Calceolaria pinnifolia [110] and structurally close lecheronol A **82** isolated from Sapium haematospermum) [111], as well as by 6-hydroxyculactone **83**, the metabolite of

Scheme 36.

Melica volkensii [75] (MIC = 4 μ g/mL). With the same value of MIC, ugandensidial **84** (from Warbugia ugandensis) inhibits the growth of M. aurum and M. phlei [75]. Diterpenes diaportheines A **85a** and B **85b** were isolated from fungus Diaporthe sp. Compound **85b** exhibits the antituberculosis activity with respect to M. tuberculosis H37Ra (MIC = $3.1 \,\mu$ g/mL) being cytotoxic, whereas compound **85a** is much less active and cytotoxic (MIC = $200 \,\mu$ g/mL) [112]. These data indicate an important role of the carbonyl group in the antituberculosis activity. A metabolite of African tree Combretum imberbe, traditionally used in folk medicine, is imberbic

acid 86 which exhibits the activity with respect to M. fortuitum (MIC = 1.56 μ g/mL) [113].

Chemical modifying the ursolic acid (via entering the fragment of substituted cinnamic acid into the position C-3) causes a four-fold increase in antituberculosis activity of this type of compounds **87a,b** (MIC = 3.13 µg/mL with respect to M. tuberculosis H37Ra, for the ursolic acid MIC = 12.5 µg/mL) [114]. Triterpene **88** isolated from the leaves of Elateriospermum tapos is active with respect to M. tuberculosis H37Ra (MIC = 3.13 µg/mL, for isoniazid and kanamycin the MIC value amounts to 0.5 and 1.25 µg/mL, respectively) [115]. Aegicerin **89a**

and protoprimulagenin A 89b were isolated from Aegiceras spp., Embelia schimperi and Peruvian plant Clavija procera. Aegicerin 89a was tested in 37 different tuberculosis strains (MIC = $1.6-3.1 \,\mu g/mL$ with respect to one strain H37Rv, 21 drug-susceptible clinical strains, two clinical isolates drug-resistant with respect to isoniazid and 13 clinical strains with MDRTB). The absence of the activity inherent in protoprimulagenin A 89b (MIC 200 µg/mL) confirms the hypothesis that the presence of the carbonyl groups, as in the case of compound 85a and 85b represents a determining factor for the antituberculosis activity. For the first time the triterpene of the oleane type demonstrates such a uniformly high activity with respect to a wide range of either drug-susceptible, and drug-resistant strains. Unfortunately, its excellent antituberculosis activity concerning a large number of strains with MDRTB (for the comparison: MIC values for isoniazid range within 4-32 µg/mL, MIC values for rifampicin ranging

within $2-16 \,\mu\text{g/mL}$) has not found any practical application until now [116] (Scheme 37).

Steroids

Saringosterol isolated from brown seaweed Sargassum ringgoldianum and Lessonia nigrescens in the form of mixture (1:1) of 24R isomer 90a and 24S isomer 90b, inhibits the growth of M. tuberculosis H37Rv (MIC = 0.25 µg/mL) exhibiting, in addition, a low cytotoxicity level. These pure isomers exhibit a different activity level (the MIC value for 24R isomer is equal to 0.125 µg/mL, the MIC value for 24S isomer being equal to $1 \mu g/mL$) [117]. From the extract of Morinda citrifolia (Rubiaceae) traditionally used in the Philippine folk medicine for treating various tuberculosis and respiratory diseases, lipids were isolated those inhibit the growth of M. tuberculosis H37Rv. The most intense activity is exhibited by the mixture of compounds 91 and 92 (MIC $\leq 2.0 \,\mu\text{g/mL}$

Scheme 37.

Scheme 38.

in the ratio of 2:1) as well as endoperoxide 93 (MIC = $2.5 \,\mu\text{g/mL}$) [118]. The derivatives of sterine 94a-e isolated from the extract of Argentinean plant *Ruprechtia triflora* are active with respect to *M. tuberculosis* (MIC = $2-4 \,\mu\text{g/mL}$) [110]. Synthetic analogues of $5(6\rightarrow7)$ abeo-sterol, the metabolite of Caribbean marine sponge *Svenzea zeai*, compound 95a,b, well inhibits the growth of *M. tuberculosis* H37Rv ATCC 27294 (MIC = $3.8 \, \text{and} \, 3.95 \,\mu\text{g/mL}$, respectively), but exhibit moderate cytotoxicity [119] (Scheme 38).

CONCLUSION

The analysis of the material of the review allows one to draw a conclusion concerning the fact that are there are no papers available from the literature devoted to the synthesis of substances those represent the structures of naturally occurring metabolite of antituberculosis action linked together with synthetic mycostatics.

Meanwhile, the studies in the field of design and synthesis of such agents are performed in the course of the development of medicinal

preparations with the most different scopes of medical use.

One could hope that the synthesis of potential mycostatic agents with mixed structure types would result in expected success and would allow one to reveal highly promising substances.

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